



Atal Medical & Research University, H.P.

(A State Govt. University)

(SLBS Govt. Medical College & Hospital Campus, Ner Chowk, Mandi, H.P.)

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Answer keys
for CET (Non-Medical) for B. Pharmacy
(Ayurveda) Entrance test
Paper Code-2505
held on 11th June, 2025
Evening Shift (02:00 PM to 04:00 PM)

If a candidate does not satisfy with the key answers, he/she can make representation with documentary proof to the COE, Atal Medical and Research University, HP, at Nerchowk, Distt. Mandi, HP-175008 through email on coemedicaluniv.mandi@gmail.com up to 13.06.2025 by 05:00 PM and thereafter, no representation will be entertained under any circumstances.

-Sd/-

(Dr. Parveen Kumar Sharma)

Controller of Examinations

Atal Medical & Research University, HP,
Ner Chowk, Distt. Mandi.

(To be filled by the candidate)

Booklet S. No. _____

Roll No. (In figures) _____

Roll No. (In words) _____

Name of the Candidate : _____

Father's Name : _____

Center of Examination : _____

Total No. of Questions : 100

Paper Code- 2505

Time Allowed : 2 Hours

Maximum Marks : 100

IMPORTANT NOTE:

- (i) OMR Answer Sheet will be supplied by the Examination Centre Superintendent for answering the questions.
- (ii) Use **Blue/Black Ink/ Ball Pen only**, to darken the appropriate circle in the OMR Answer Sheet.
- (iii) Darken one circle deeply for each question in the OMR Answer Sheet, as faintly darkened circle might be rejected by the Optical Scanner.
- (iv) Darkening of more than one circle shall be rejected by the scanner. **Over-writing, cutting, erasing or use of White Fluid is not allowed.**
- (v) Before dealing with the question paper, fill-up the required information with Blue/Black Ball Pen correctly both in the Question Booklet and in the OMR Answer Sheet.
- (vi) Do not fold the OMR Answer Sheet nor put any mark here and there to avoid rejection by the Optical Scanner.
- (vii) Write Roll. No. carefully on the OMR Answer Sheet and darken the appropriate circle properly.
- (viii) **Each question carries One Mark. There will be No Negative Marking.**
- (ix) **Use of Calculator is not allowed.**
- (x) **No over-writing, cutting, erasing or use of white fluid is allowed.**
- (xi) Rough work be done on the sheet(s) at the end.
- (xii) **MOBILE TELEPHONES (EVEN ON SWITCH-OFF MODE) AND SUCH OTHER ELECTRONIC DEVICES ARE NOT ALLOWED INSIDE THE EXAMINATION HALL.**
- (xiii) The question paper-booklet will be retained by the candidate after the entrance test is over.
- (xiv) Question No 1 - 40 is Math, 41 – 70 is Chemistry and 71 - 100 is Physics.

Paper Code- 2505

Rough Page

AMRC

Math

Q1. If A and B are two disjoint sets then

$A \cap B$ is

- a) Universal set
- b) empty set
- c) A
- d) None

Ans : B

Q2. If $n_{c_6} = n_{c_7}$ then n is

- a) 3
- b) 13
- c) 5
- d) 4

Ans : B

Q3. $\cos\left(\frac{\pi}{2} + x\right) = ?$

- a) $\sin x$
- b) $\cos x$
- c) $-\sin x$
- d) $\tan x$

Ans : C

Q4. Principal value of $\tan^{-1} 1$ is

- a) $\frac{\pi}{2}$
- b) $\frac{\pi}{4}$
- c) $-\frac{\pi}{2}$
- d) $-\frac{\pi}{4}$

Ans : B

Q5. The value of $\hat{i} \cdot (\hat{j} \times \hat{k}) + \hat{j} \cdot (\hat{i} \times \hat{k}) + \hat{k} \cdot (\hat{i} \times \hat{j})$ is

- a) 0
- b) -1
- c) 1
- d) 3

Ans : C

Q6. If $P(A) = \frac{5}{6}$

then $P(\bar{A})$ (Probability not A)

- a) 0
- b) $\frac{7}{6}$
- c) 2
- d) $\frac{1}{6}$

Ans : D

Q7. If $P(A) = \frac{7}{13}$ $P(B) = \frac{9}{13}$

and $P(A \cap B) = \frac{4}{13}$

a) $\frac{5}{9}$

b) $\frac{9}{4}$

c) $\frac{9}{7}$

d) $\frac{7}{7}$

Ans : A

Q8. $\int \cos x \, dx = ?$

a) $\sin x$

b) $\tan x$

c) $-\sin x$

d) $\cos x$

Ans : A

Q9. Integration of $\cos^2 x$ is :

a) $-\cos x + c$

b) $-\sin x + c$

c) $-\cot x + c$

d) None

Ans : C

Q10. Value of i (iota)

a) 0

b) $\sqrt{-1}$

c) -1

d) 1

Ans : B

Q11. Value of $i^8 = ?$

a) 1

b) -1

c) $\sqrt{-1}$

d) 4

Ans : A

Q12. If $n=7$ and $r=4$ then value of n_{c_r} is

a) 45

b) 50

c) 35

d) 100

Ans : C

Q13. The series $a, ar, ar^2, \dots$ is

- a) AP
- b) GP
- c) HP
- d) None of these

Ans : B

Q16. Vertex of Parabola $y^2=16x$ is

- a) (-9,0)
- b) (0,4)
- c) (0,0)
- d) (1,1)

Ans : C

Q14. 10th term of GP 3,9,27,81is

- a) 3^5
- b) 10^3
- c) 3^{10}
- d) 3^3

Ans : C

Q17. The value of $\lim_{x \rightarrow 0} \frac{\tan x}{x} = ?$

- a) X
- b) 1
- c) 0
- d) ∞

Ans : B

Q15. General equation of ellipse is

- a) $ax + by + c = 0$
- b) $y^2 = 4ax$
- c) $y = mx + c$
- d) $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$

Ans : D

Q18. Derivative of $\tan x$ is

- a) $\operatorname{Cosec}^2 x$
- b) $\sec x \tan x$
- c) $\sec^2 x$
- d) $\cos x$

Ans : C

Q19. Slope of Line $-3x+2y=7$ is

- a) $\frac{3}{2}$
- b) $\frac{2}{3}$
- c) $\frac{7}{2}$
- d) All of These

Ans : A

Q20. Find $\frac{dy}{dx}$ if $x = A \cos \theta$ $Y = B \cos \theta$

- a) B/A
- b) A/Bs
- c) 1
- d) None of These

Ans : A

Q21. Which is true for inverse of Matrix A and B

- a) $A^{-1} \cdot B^{-1} = B \cdot A$
- b) $AB = BA$
- c) $(AB)^{-1} = B^{-1} A^{-1}$
- d) None

Ans : C

Q22. The product of two Matrix $A \cdot B = I$

Then which is true about A and B

- a) $A=B$
- b) $A=B^{-1}$
- c) $AB=(AB)^{-1}$
- d) None of these

Ans : B

Q23. $\int \sec x \cdot (\sec x + \tan x) dx = ?$

- a) $\tan x + \sec x + C$
- b) $\tan x$
- c) $\tan^2 x$
- d) 1

Ans : A

Q24. Distance between two point $A(x_1, y_1)$ and $B(x_2, y_2)$

- a) $\sqrt{(x_2 - x_1)^2 + (y_2 - y_1)^2}$
- b) $\sqrt{x_1^2 + y_1^2}$
- c) $\sqrt{x_1^2 \times y_1^2}$
- d) xy

Ans : A

Q25. Which is Ir-ration Numbers

- a) π
- b) $\sqrt{2}$
- c) $\sqrt{\frac{1}{3}}$
- d) All of these

Ans : D

Q26. $\frac{\tan x + \tan y}{1 - \tan x \tan y}$ is equal to

- a) $\tan(x-y)$
- b) $\sin(x-y)$
- c) $\tan(x+y)$
- d) $\sin(x+y)$

Ans : C

Q27. Total No's of term in Binomial expansion $(x+y)^6$

- a) 6
- b) 7
- c) 8
- d) 9

Ans : B

Q28. $(100+1)^4$ fifth term of this Binomial expansion accordingly by term

- a) 1
- b) 100
- c) 1000000
- d) 1000

Ans : A

Q29. Which Series is G.P

- a) 2,4,6,8 _ _ _ _
- b) 1,3,5,7 _ _ _ _
- c) 1,2,3,4 _ _ _ _
- d) 5,25,625,3125 _ _ _ _

Ans : D

Q30. Geometric mean of a and b is

- a) $\frac{a+b}{2}$
- b) $\sqrt{ab}$
- c) $\frac{a-b}{2}$
- d) $(ab)^2$

Ans : B

Q31. $Y = \tan 45^\circ$ Then $\frac{dy}{dx}$ is

- a) $\sec^2 45^\circ$
- b) 1
- c) $\tan 45^\circ$
- d) 0

Ans : D

Q34. Volume of sphere of radius P is

- a) P^3
- b) $\frac{2}{3}\pi P^3$
- c) $2\pi P^3$
- d) $\frac{4}{3}\pi P^3$

Ans : D

Q32. $Y = f(x)$ is one - one. If $x_1 = x_2$ Then

- a) $f(x_1) = f(x_2)$
- b) $f(x_1) \neq f(x_2)$
- c) $f(x_1) f(x_2) = 0$
- d) None of these

Ans : A

Q35. Equation of circle which pass through origin and radius 'r' is

- a) $X^2 + y^2 = r^2$
- b) $X^2 + y^2 = 0$
- c) $x = y$
- d) $xy = 0$

Ans : A

Q33. If $\vec{A}$ and $\vec{B}$ are perpendicular then

- a) $\vec{A} \cdot \vec{B} = 0$
- b) $\vec{A} + \vec{B} = 0$
- c) $\vec{A} \times \vec{B} = 0$
- d) $\frac{\vec{A}}{\vec{B}} = 0$

Ans : A

Q36. If roots of Quadratic equation $ax^2 + bx + c = 0$ is equal then

- a) $b^2 - 4ac = 0$
- b) $b^2 - 4ac > 0$
- c) $b^2 - 4ac < 0$
- d) $b^2 - 4ac = 1$

Ans : A

Q 37. If $f(x) = 6x^2 - 9$ then $f(-2)$ is

- a) 15
- b) 6
- c) 24
- d) 23

Ans : A

Q 38. If set A has 3 elements then A^2 has

- a) 6, elements
- b) 9, elements
- c) 0, elements
- d) 100, elements

Ans : B

Q39. $\lim_{x \rightarrow 7} \frac{x^2 - 49}{x - 7}$ is equal to

- a) 14
- b) 7
- c) 21
- d) 49

Ans : A

Q40. Determinants of Matrix A =

$$\begin{bmatrix} 2 & 4 \\ -5 & -1 \end{bmatrix} \text{ is}$$

- a) 11
- b) 19
- c) 18
- d) 0

Ans : C

Chemistry

Q41. Which metal is used in Ullmann reaction?

- (a) Na
- (b) Mg
- (c) Cu
- (d) Zn

Answer: C

Q42. Fehling A and Fehling B are :

- (a) CuSO_4 soln. + AgNO_3
- (b) CuSO_4 and alkaline solution of Sodium tartarate
- (c) Both a & b
- (d) $\text{CuSO}_4 + \text{CaSO}_4$

Answer : B

Q43. Which of the following is tertiary amine?

- (a) Triethylamine
- (b) N-methyl aniline
- (c) 1-methylcyclohexyl amine
- (d) none of the above

Answer : A

Q44. Aspirin is :

- (a) Acetyl salicylic acid
- (b) Phenyl salicylate
- (c) Acetyl salicylate
- (d) Methyl salicylic acid

Answer : A

Q45. In Clemmensen Reduction, Carbonyl compound is treated with

- (a) Zinc amalgam with HNO_3
- (b) Sodium amalgam + HCl
- (c) Zinc amalgam + HCl
- (d) Sodium amalgam + HNO_3

Answer : C

Q46. Vitamin is soluble in water?

- (a) Vitamin D
- (b) Vitamin K
- (c) Vitamin E
- (d) Vitamin B

Answer : D

Q47. $\text{CF}_2=\text{CF}_2$ is a monomer of which of the following polymer?

- (a) Teflon
- (b) Nylon-6
- (c) Glyptal
- (d) Buna-S

Answer : A

Q48. Pain relieving drugs are called

- (a) Antipyretics
- (b) Analgesics
- (c) Antibiotics
- (d) Antiseptics

Answer : B

Q49. What will be the X in the following reaction: Phenol + Zn dust + heat = X

- (a) Acetic Acid
- (b) Ethanol
- (c) Methanol
- (d) Benzene

Answer : D

Q50. Composition of a compound is C=26%, H=8% and rest is oxygen. Then % of Oxygen will be

- (a) 16%
- (b) 66%
- (c) 100%
- (d) 34%

Answer : B

Q51. Match column X and Y

Column X

- i. Acetic acid
- ii. Sulphuric acid
- iii. Formaldehyde
- iv. Lime stone

Column Y

- a. CaCO_3
- b. CH_3COOH
- c. HCHO
- d. H_2SO_4

- (a) i-a, ii-b, iii-c, iv-d
- (b) i-b, ii-a, iii-d, iv-c

- (c) i-b, ii-d, iii-c, iv-a
- (d) none of the above

Answer : C

Q52. Lindlar's catalyst is :

- (a) Pt in ethanol
- (b) Pd with BaSO_4
- (c) Ni in ethanol
- (d) Na in liq. NH_3

Answer : B

Q53. What is Lucas Reagent

- (a) A mixture of $\text{HCl} + \text{ZnCl}_2$
- (b) A mixture of $\text{HCl} + \text{CaCl}_2$
- (c) A mixture of $\text{HNO}_3 + \text{ZnCl}_2$
- (d) None of the above

Answer : A

Q54. $\text{C}_6\text{H}_5\text{NH}_2 \xrightarrow[\text{HCl}]{\text{NaNO}_2}$?

Identify ?

- (a) benzene diazonium Chloride.
- (b) $\text{C}_6\text{H}_5\text{NH}^{+3}$
- (c) $\text{C}_6\text{H}_5\text{Cl}$
- (d) None of the above.

Answer : A

Q55. Number of Significant figures in 6.626 are

- (a) 3
- (b) 5
- (c) 4
- (d) 2

Answer : C

Q56. Benzoic acid is

- (a) $\text{C}_6\text{H}_5\text{CHO}$
- (b) $\text{C}_6\text{H}_5\text{COOH}$
- (c) CH_3COOH
- (d) CH_3CHO

Answer : B

Q57. The hybridisation of 'S' in SF_6

- (a) sp^3d^2
- (b) sp^2
- (c) sp^3d
- (d) sp^3d^3

Answer : A

Q58. Which concentration term is unitless?

- (a) Molarity
- (b) Molality
- (c) Mole fraction
- (d) Formality

Answer : C

Q59. The Highest magnetic moment by a metal ion with outer Electronic configuration :

- (a) 3d^2
- (b) 3d^7
- (c) 3d^9
- (d) 3d^5

Answer : D

Q60. If 'n' is '2' then values of 'l' quantum no. can be

- (a) 0, 1, 2
- (b) 0, 2
- (c) 0, 1, 3
- (d) 0, 1

Answer : D

Q61. The method used to remove temporary hardness of water is

- (a) Synthetic resins method
- (b) Calgon's method
- (c) Clark's method
- (d) Ion-exchange method

Answer : C

Q62. Which of the following pair is not isoelectronic in nature

- (a) Na^+ , Mg^{+2}
- (b) Na^+ , Al^{3+}
- (c) Mg^{2+} , Al^{3+}
- (d) Na^+ , K^+

Answer : D

Q63. Syn gas is mixture of

- (a) $\text{CO}_2 + \text{H}_2$
- (b) $\text{CO} + \text{H}_2$
- (c) $\text{CO} + \text{CO}_2$
- (d) $\text{CO} + \text{N}_2$

Answer : B

Q64. The main contributor of acid rain is

- (a) Sulphuric acid
- (b) Calcium oxide
- (c) Calcium carbonate
- (d) PAN

Answer : A

Q65. Chlorobenzene when treated with sodium in dry ether gives diphenyl, the reaction is ...

- (a) Fittig reaction
- (b) Wurtz reaction
- (c) Sandmeyer reaction
- (d) Coupling reaction.

Answer : A

Q66. Which metal is present in B_{12} Vitamin?

- (a) Fe
- (b) Co
- (c) Mg
- (d) Pt

Answer : B

Q67. Which of the following orbital is nearest to the nucleus?

- (a) 4f
- (b) 5d
- (c) 4s
- (d) 7p

Answer : C

Q70. Which of the following is not a colligative property?

- (a) Elevation in B.p t
- (b) Depression in freezing point
- (c) Osmotic pressure
- (d) Lowering in vapour pressure

Answer : D

Q68. What is the formula of Borax?

- (a) $\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$
- (b) $\text{Na}_2\text{B}_2\text{O}_7$
- (c) $\text{NaBO}_2 + \text{B}_2\text{O}_3$
- (d) B_2O_3

Answer : A

Physics

Q71. Whenever a wave enters from one medium to another, its

- a) Velocity changes
- b) frequency changes
- c) frequency doesn't change
- d) wavelength remains constant

ANS c

Q69. Among the following ions which one has the highest paramagnetism?

- (a) $[\text{Cr}(\text{H}_2\text{O})_6]^{3+}$
- (b) $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$
- (c) $[\text{Cu}(\text{H}_2\text{O})_6]^{2+}$
- (d) $[\text{Zn}(\text{H}_2\text{O})_2]^{2+}$

Answer : B

Q72. A particle moves in a circular path with a uniform speed, its motion is

- a) periodic
- b) oscillatory
- c) simple harmonic
- d) angular simple harmonic

Ans (a)

Q73. What is the number of degrees of freedom for an ideal diatomic molecule at ordinary temperature?

- a) 3
- b) 5
- c) 7
- d) 6

Ans (b)

Q74. First law of thermodynamics is the law of conversion of

- a) mass
- b) linear momentum
- c) energy
- d) angular momentum

Ans c

Q75. S.I. unit of energy is

- a) joule
- b) calorie
- c) kilo calorie
- d) none

Ans (a)

Q76. Plants get water through the roots because of

- a) Capillarity
- b) Viscosity
- c) Gravity
- d) elasticity

Ans-a

Q77. Dimensional formula stress is same as that of

- a) force
- b) impulse
- c) pressure
- d) strain

Ans-c

Q78. When you move from equator to pole, the value of mass of the body

- a) increases
- b) decreases
- c) remains the same
- d) increases then decreases

Ans-c

Q79. Torque acting on a body is the rotational analogue

- (a) mass
- b) force
- c) velocity
- d) K.E.

Ans-b

Q82. which of the following quantities is vector

- a) work
- b) K.E.
- c) power
- d) Angular Moment

Ans-d

Q80. A light and heavy body have equal momentum which one has greater Kinetic energy

- a) A light body
- b) heavy body
- c) both have equal K.E.
- d) none

Ans-a

Q83. If a body travels with constant acceleration which of the following quantities remains Constant

- a) Displacement
- b) Velocity
- c) Time
- d) None

Ans-d

Q81. Newton's first law of motion describes the

- a) inertia
- b) energy
- c) work
- d) momentum of inertia

Ans-a

Q84. Electron volt is a unit of

- a) Potential difference
- b) Electric current
- c) electronic charge
- d) Energy

Ans-d

Q85. Which of the following physical quantity is dimensionless?

- a) Angle
- b) specific gravity
- c) Strain
- d) all of these

Ans-d

Q88. Which spectral series of hydrogen lie in UV region?

- a) Paschen
- b) Balmer
- c) Lyman
- d) Brackett

ANS-C

Q86. Majority charge carriers in p-type semiconductor are

- a) Holes
- b) Electrons
- c) Holes and Electrons
- d) None

Ans – a

Q89. The mass of photon at rest is

- a) 1
- b) Infinite
- c) Zero
- d) None

ANS-C

Q87. An atom bomb works on the principle of

- a) Alpha-Decay
- b) Beta decay
- c) nuclear fission
- d) Nuclear fusion

Ans. C

Q90. Mirage is due to which of the following

- a) Diffraction
- b) Reflection
- c) Refraction
- d) Total Internal reflection

ANS-d

Q91. Focal length of the plane mirror is

- a) Negative
- b) positive
- c) Zero
- d) infinite

ANS-d

Q92. Which of the following are not eM waves?

- a) Cosmic rays
- b) r-rays
- c) β -rays
- d) x-rays

Ans-c

Q93. The direction of Induced Emf is given by

- a) Fleming's Law
- b) Ampere's Law
- c) Lenz's Law
- d) Lorentz's Law

Ans-c

Q94. A uniform magnetic field is obtained in a

- a) bar magnet
- b) horse shoe magnet
- c) circular coil carrying current
- d) none

Ans a

Q95. A charged particle enters a magnetic field at an angle 90° with magnetic field. The path of the particle will be

- b) a helix
- c) an ellipse
- c) a circle
- d) a straight line.

Ans c

Q96. The unit of electric power is

- a) Volt
- b) Joule
- c) Watt
- d) kwh

Ans-c

Q97. In which of the following forms is the energy stored in a capacitor

- a) charge
- b) Potential
- c) Capacitance
- d) electric field

Ans-d

Q99. The net charge on a conductor is

- a) Positive
- b) negative
- c) Varying
- d) zero

Ans-d

Q98. Which of the following is the best insulator

- a) Carbon
- b) Paper
- c) Graphite
- d) Ebonite

Ans-d

Q100. The force per unit charge is known as

- a) Electric flux
- b) Electric field
- c) Electric potential
- d) electric current

Ans-b